

**ASYMMETRIC SYNTHESIS OF PYRROLIZIDINONES BY RADICAL CYCLIZATION
OF N-ALLYLIC PYROGLUTAMATES ^{1a}**

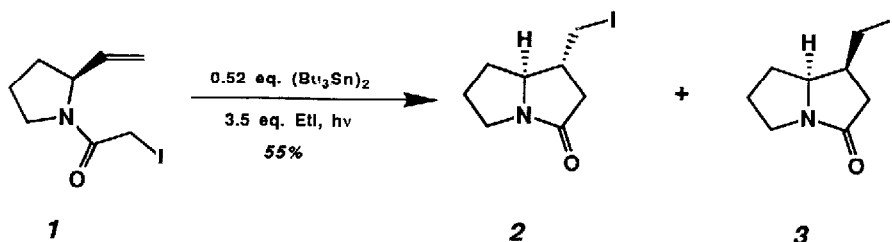
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Abstract: Ethyl S-pyroglutamate is converted to N-allylic-S C₅-iodomethyl-2-pyrrolidinone. Facile radical cyclization on treatment with AIBN and tributyltin hydride in refluxing benzene gives good yields of the C₆ substituted pyrrolizidin-2-one with excellent diastereoselectivity.

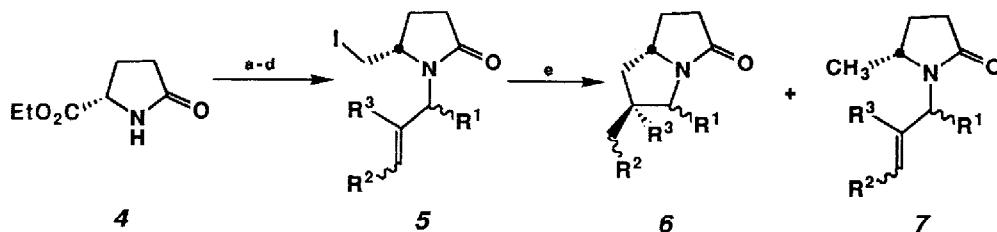
Radical cyclization of haloalkenes is an important tool for construction of carbocyclic rings in natural product synthesis.² An intermediate alkenyl radical is generated by treatment of haloalkenes with a radical initiator such as azo-bis-isobutyronitrile (AIBN) or by photochemical methods. Tributyltin hydride (Bu₃SnH) is usually added to transfer an hydrogen atom to the cyclized radical product. 1-Azabicyclo[m.n.0]alkanes are an important class of naturally occurring alkaloids of great interest to synthetic chemists.³ The pyrrolizidine alkaloids (m = n = 1), for example, possess important biological activities ranging from insect attractants or defensive substances^{3b} to anticholinergic^{3b} or antineoplastic activity.^{3b} Most of the synthetic routes to 1-azabicyclo[m.n.0] alkaloids rely on nucleophilic ring closures using Dieckmann or Aldol condensations.³ A few strategies use acid catalyzed ring closure routes. 3 + 2 Cycloadditions with nitron derivatives also lead to efficient syntheses of these alkaloids.³ Radical cyclization has been applied sparingly to alkaloid syntheses but pyrrolizidines are well suited to this strategy.

Radical cyclization of heteroatom containing substrates is known for many alkenyl halides^{2,4} and two radical cyclization strategies have been used to construct pyrrolizidines. In the first Hart cyclized 5-thiophenyl-N-alkenyl-2-pyrrolidinones with AIBN and Bu₃SnH,⁵ but produced mixtures of pyrrolizidines and indolizidines with poor diastereoselectivity and no asymmetric induction. More recently, Hart reported an asymmetric synthesis of (-)-swainsonine by radical cyclization of an asymmetric N-alkynyl-5-thiophenyl precursor.⁶ The chiral succinimide precursor was prepared from tartaric acid, and radical cyclization gave excellent diastereoselectivity and a 70% yield of the desired indolizidinone products. In a recent prolinol based strategy,⁷ Livinghouse prepared **1** via Wittig reaction with a BOC-protected prolinol. ⁸ Cyclization with Bu₃SnH/AIBN gave poor yields of the pyrrolizidine, but radical generation with *n*-(Bu₃Sn)₂/hν gave a 65:1 mixture of **2** and **3** in 58% yield.⁷ Pyrrolizidin-2-one **2** was converted to (-)-trachelanthamidine in two steps.⁷ Other non-radical prolinol strategies, involving intramolecular cyclization via carbanionic intermediates have been reported for the asymmetric synthesis of 'izidine' alkaloids such as septicine.⁹



S-Ethyl pyroglutamate (**4**) is prepared in high yield by reaction of glutamic acid with thionyl chloride and heating in ethanol¹⁰. Alkylation of **4** with a variety of allylic halides required 1.5 equivalents of powdered KOH, 20 mol% of Bu₄NBr as a phase transfer catalyst (analogous to the work of Takahata)¹¹ in THF but gave yields of only 5-10%. Sonication (ultrasonic cleaning bath) during the reaction led to good to excellent yields of the desired *N*-substituted lactams (81% with allyl bromide). The alkylation reaction did not racemize the C₅ hydrogen.

Standard LiBH₄ reduction of the ester gave only 0-10% yield of the requisite hydroxy-methyl lactam. Reduction with lithium aluminum hydride on anhydrous silica gel,¹² however, gave 70-85% of the alcohol.¹³ Conversion of the hydroxymethyl moiety to the mesylate (one equivalent of CH₃SO₂Cl and one equivalent of alcohol in CH₂Cl₂ were treated with 1.1 equivalents of triethylamine at -78°)¹⁴ allowed formation of the iodomethyl derivative by Finkelstein exchange (10 equivalent of sodium iodide in refluxing acetone for 1 hour). In sharp contrast to the cyclization of **1**, treatment of one equivalent of **5** with 2 equivalents of Bu₃SnH (5-10 mol% AIBN in refluxing benzene [5-25 mM], 4-12 hours) gave good yields of cyclized product, **6** (see Table 1). Iodolactam **5a** gave **6a** in 70% yield.¹⁵ In all cases except **5d**, **7** (resulting from hydrogen transfer to the intermediate radical) was the only other product. Identification of **6** was straightforward via NMR and capillary GC/Mass spectral analysis.



(a) R²CH=C(R³)CHR¹Br / THF/sonication/Bu₄NBr/KOH (b) LiAlH₄/SiO₂ (c) CH₃SO₂Cl/NEt₃
 (d) NaI/acetone (e) AIBN/ Bu₃SnH / PhH/reflux

As an example, asymmetric induction in **6a** was confirmed ($[\alpha]_{25}^D = +20.8$ [EtOH, c. 0.0288 g/mL]) and NMR analysis clearly showed the C₆ methyl group to be *cis*- to the hydrogen at C_{7a} (*exo*-methyl). The proton at C_{7a} appears at 4.04 ppm; the methyl group at 1.09 ppm; the *exo*-proton at C₅ at 3.81 ppm and the *endo*-hydrogen at C₅ at 2.42 ppm. The COSY spectrum shows strong coupling of the methyl group and C₆ hydrogen, which appears at 2.33 ppm, but no coupling of the C_{7a} and C₆ hydrogens. Enhanced long range coupling (*W*-type) in the COSY for the C_{7a} hydrogen and the *exo*-C₅ hydrogen is apparent. Similar long range coupling for the C₆-C_{7a} hydrogens was absent, strongly suggesting the methyl

group is *exo* (*cis*-) to the C_{7a} hydrogen. There was no enhancement for these signals in the NOESY spectrum.

Similar *exo*-selectivity was observed in the prolinol based system.² Models show a steric interaction of the alkenyl moiety with the pyrrolidinone ring in the *endo*-transition state which is absent in the *exo*- transition state. No significant interaction of the C_{7a} hydrogen was apparent in any transition state leading to cyclized product. Although the energy differential between these rotamers may be small, the diastereoselectivity for the *exo*-transition state was clear. A single product was detected in the ¹H NMR and by GC/MS analysis. Further analysis by HPLC (C₁₈-reverse phase, acetonitrile) showed less than 1% of a peak which *may* be the *endo*- diastereomer. Analysis by GC/MS showed that no indolizidin-2-one was formed in the cyclization of **5a**.¹⁶ The diastereoselectivity of this reaction appears comparable to that of the prolinol-based cyclization.

Indolactams **5b** and **5c** were cyclized to **6b** and **6c**, respectively, under identical conditions (see Table 1) with *exo*-selectivity. Cyclization of **5d** introduced the problem of selectivity between the methylene and methyl moieties. A 30:70 mixture of **6d** and 6-methylindolizidin-2-one was isolated, the remainder of the product being **7d** although this was greatly diminished at concentrations of less than 5 mM. The product distribution suggests a slight preference for the rotamer with the methylene group *endo*-. The steric encumbrance inherent to an *endo*- methyl or methylene probably leads to a rotamer in which the terminal methylene carbon is exposed to attack, giving the six-membered ring. We did *not* observe indolizidin-2-one products in any other case¹⁶ but **7** was the by-product in reactions of **5a-f**.

Table 1. Radical Cyclization of N-Allylic-5-iodomethyl-2-pyrrolidinones.

<u>5/6^a</u>	<u>R¹</u>	<u>R²</u>	<u>R²</u>	<u>Time (hr)</u>	<u>Conc (mM)</u>	<u>% 6^{a,b}</u>
a	H	H	H	4	10	70
b	H	Ph	H	3	25	72
c	H	Me	H	12	3	52
d	H	H	Me	12	5	96 ^c
e	Me	H	H	12	4.3	54
f	-CH ₂ CH ₂ CH ₂ -	H	H	12	6	58

^a Satisfactory analysis for all new compounds ^b remaining product is **7**

^c a 70:30 mixture of 6-methylindolizidin-2-one and **6d**

Substitution at the position α to the nitrogen, R¹ in **5e** and **5f** leads to the expected mixture of diastereomers. The methyl group at C₆ is *exo*- but the 70:30 diastereomeric mixture in **5e** is reflected in the cyclized product, **6e**. The conformational bias for the *equatorial* isomer in **5f** leads to some selectivity for the *trans*-fused B-C ring in **6e**. Only one product was easily discernible in the ¹H NMR but HPLC shows a 30:1 mixture of diastereomers for **6f**.

The facility of the pyroglutamate based radical cyclization under milder conditions than the prolinol based cyclization is obvious from these results. The high preference for *exo*-substitution during the cyclization is apparent. COSY and NOESY are powerful tools for

determining the relative stereochemistry of substituents in these pyrrolizidinones. Modification of this pyroglutamate strategy to a C₅-alkenyl *N*-iodoethyl derivative is underway and will allow the asymmetric synthesis of a variety of naturally occurring pyrrolizidine alkaloids.

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- Addition of 1.30 g (4.90 mmol) of **5a**, 0.45 g (9.80 mmol) of Bu₃SnH and 0.05 g (0.3 mmol) of AIBN in 500 mL of dry benzene was followed by reflux for 4 hours. The solution was cooled, concentrated *in vacuo* and chromatography (silica gel/ether) gave 0.48 g (3.45 mmol, **70%**) of **6a**: ¹H NMR (CDCl₃) δ 1.09 (d, 3H), 1.61 (m, 4H), 2.33 (m, 1H), 2.42 (m, 1H), 2.61 (m, 2H), 3.81 (dd, 1H) and 4.04 ppm (m, 1H); ¹³C NMR (CDCl₃) : δ 19.6 (q), 28.3 (t), 34.5 (t), 34.9 (t), 39.7 (t), 49.3 (t), 60.2 (d) and 175.3 ppm (s); Mass spectrum (m/z, rel. intensity): 139 (P,50), 124 (3), 110 (2), 97 (100), 84 (12), 69 (50), 55 (40) and 41 (45).
- Indolizidin-2-ones show a pronounced P-1 peak in the mass spectrum. This peak was missing in all pyrrolizidin-2-ones, allowing differentiation by GC/MS.

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